



## Ferrous Metals



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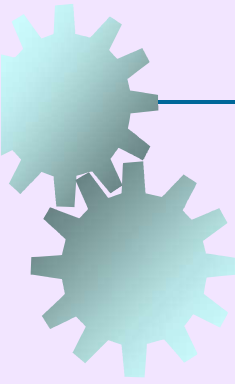
Based on **iron**, one of the oldest metals known to man

- Ferrous metals of engineering importance are **alloys of iron and carbon**
- These alloys divide into two major groups:
  - **Steel**
  - **Cast iron**
- Together, they constitute approximately 85% of the metal tonnage in the United States



## Steel and Cast Iron

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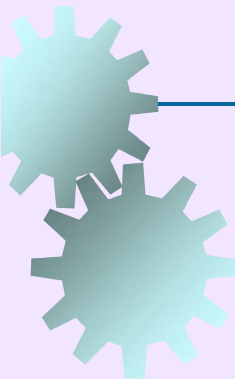


What is the difference between steel and cast iron?!



## Steel and Cast Iron Defined

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**Steel** = an iron-carbon alloy containing from 0.02% to 2.1% carbon

**Cast iron** = an iron-carbon alloy containing from 2.1% to about 4% or 5% carbon

- Steels and cast irons can also contain other alloying elements besides carbon



# Iron-Carbon Phase Diagram

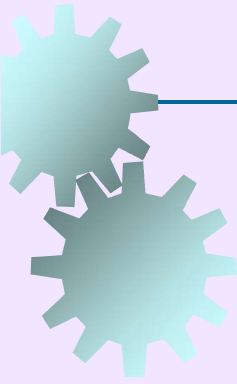
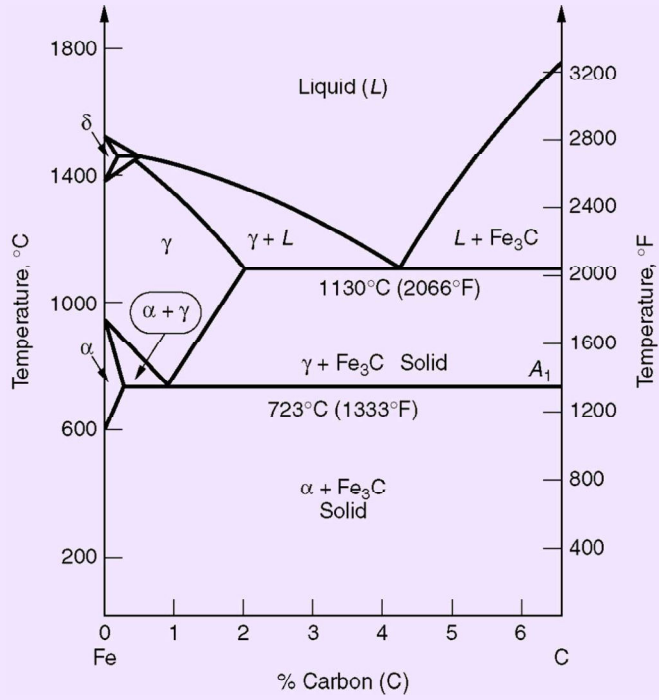


Figure 6.4 Phase diagram for iron-carbon system, up to about 6% carbon.



Steel



# Steel

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An alloy of iron containing from 0.02% and 2.11% carbon by weight

- Often includes other alloying elements: **nickel, manganese, chromium, and molybdenum**
- Steel alloys can be grouped into four categories:
  1. **Plain carbon steels**
  2. **Low alloy steels**
  3. **Stainless steels**
  4. **Tool steels**



## 1

### Plain Carbon Steels

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- Carbon is the principal alloying element, with only small amounts of other elements (about 0.5% manganese is normal)
- Strength of plain carbon steels increases with carbon content, but ductility is reduced

Carbon ↗ Strength ↗

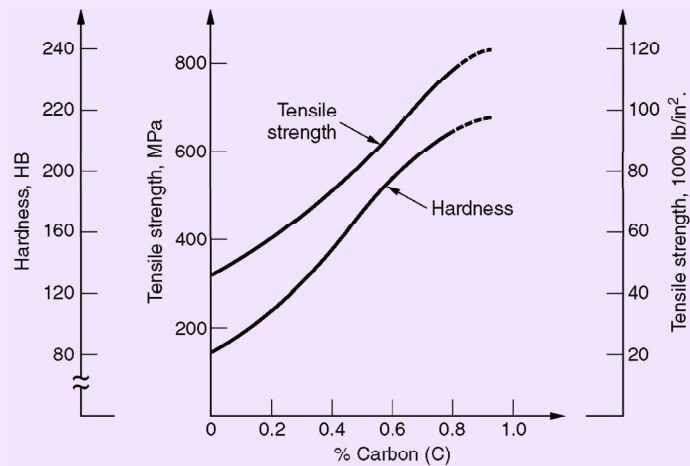
Carbon ↗ Ductility ↘

- High carbon steels can be heat treated to form martensite, making the steel very hard and strong



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Figure 6.12 Tensile strength and hardness as a function of carbon content in plain carbon steel (hot rolled).



**Hardness** is the characteristic of a solid material expressing its resistance to permanent deformation. It is expressed as Brinell hardness number or BHN or HB:

P = applied force (kgf)

D = diameter of indenter (mm)

d = diameter of indentation (mm)

$$\text{BHN} = \frac{2P}{\pi D(D - \sqrt{D^2 - d^2})}$$



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## AISI-SAE Designation Scheme

Specified by a 4-digit number system: 10XX, where **10** indicates plain carbon steel, and **XX** indicates carbon % in hundredths of percentage points

- For example, 1020 steel contains 0.20% C
- Developed by American Iron and Steel Institute (AISI) and Society of Automotive Engineers (SAE), so designation often expressed as **AISI 1020** or **SAE 1020**



# 1

## Plain Carbon Steels



### 1. Low carbon steels - contain less than 0.20% C

- Applications: automobile sheetmetal parts, plate steel for fabrication, railroad rails



### 2. Medium carbon steels - range between 0.20% and 0.50% C

- Applications: machinery components and engine parts such as crankshafts and connecting rods



### 3. High carbon steels - contain carbon in amounts greater than 0.50%

- Applications: springs, cutting tools and blades, wear-resistant parts



# 2

## Low Alloy Steels

Iron-carbon alloys that contain additional alloying elements in amounts totaling less than ~ 5% by weight



Large diameter pipeline

- Mechanical properties superior to plain carbon steels for given applications
- Higher strength, hardness, wear resistance, toughness, and more desirable combinations of these properties
- Heat treatment is often required to achieve these improved properties



## 2

# AISI-SAE Designation Scheme

AISI-SAE designation uses a 4-digit number system: YYXX, where YY indicates alloying elements, and XX indicates carbon % in hundredths of % points

- Examples:

- 13XX - Manganese steel

- 20XX - Nickel steel

- 31XX - Nickel-chrome steel

- 40XX - Molybdenum steel

- 41XX - Chrome-molybdenum steel



## 3

# Stainless Steel (SS)

Highly alloyed steels designed for corrosion resistance

- Principal alloying element is Chromium, usually greater than 15%
  - Cr forms a thin oxide film that protects surface from corrosion
- Nickel (Ni) is another alloying ingredient in certain SS to increase corrosion protection
- Carbon is used to strengthen and harden SS, but high C content reduces corrosion protection since chromium carbide forms to reduce available free Cr



Carbon ↗ Strength ↗

Carbon ↗ Corrosion protection ↘





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## Properties of Stainless Steels

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- In addition to corrosion resistance, stainless steels are noted for their combination of strength and ductility
  - While desirable in many applications, these properties generally make stainless steel difficult to work in manufacturing
- Significantly more expensive than plain C or low alloy steels



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## Types of Stainless Steel

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- Classified according to the predominant phase present at ambient temperature:
  1. **Austenitic stainless** - typical composition 18% Cr and 8% Ni
  2. **Ferritic stainless** - about 15% to 20% Cr, low C, and no Ni
  3. **Martensitic stainless** - as much as 18% Cr but no Ni, higher C content than ferritic stainless





### 3

## Designation Scheme for Stainless Steels

- Three-digit AISI numbering scheme
- First digit indicates general type, and last two digits give specific grade within type
  - Examples:
    - Type 302 – Austenitic SS  
18% Cr, 8% Ni, 2% Mn, 0.15% C
    - Type 430 – Ferritic SS  
17% Cr, 0% Ni, 1% Mn, 0.12% C
    - Type 440 – Martensitic SS  
17% Cr, 0% Ni, 1% Mn, 0.65% C



### 3

## Additional Stainless Steels

- Stainless steels developed in early 1900s
- Several additional high alloy steels have been developed and are also classified as stainless steels:



4. **Precipitation hardening stainless** - typical composition = 17% Cr and 7%Ni, with additional small amounts of alloying elements such as Al, Cu, Ti, and Mo (Aerospace applications)
5. **Duplex stainless** - mixture of austenite and ferrite in roughly equal amounts (heat exchangers, pumps)



# 4

## Tool Steels

A class of (usually) highly alloyed steels designed for use as industrial **cutting tools, dies, and molds**

- To perform in these applications, they must possess **high strength, hardness, wear resistance, and toughness under impact**
- Tool steels are heat treated



# 4

## AISI Classification of Tools Steels

- |      |   |
|------|---|
| T, M | <i>High-speed tool steels</i> - cutting tools in machining  |
| H    | <i>Hot-working tool steels</i> - hot-working dies for forging, extrusion, and die-casting                 |
| D    | <i>Cold-work tool steels</i> - cold working dies for sheetmetal pressworking, cold extrusion, and forging |
| W    | <i>Water-hardening tool steels</i> - high carbon but little else  |
| S    | <i>Shock-resistant tool steels</i> - tools needing high toughness, as in sheetmetal punching and bending  |
| P    | <i>Mold steels</i> - molds for molding plastics and rubber  |





## Cast Iron



## Cast Irons

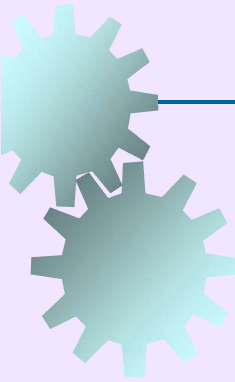
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Iron alloys containing from 2.1% to about 4% carbon and from 1% to 3% silicon

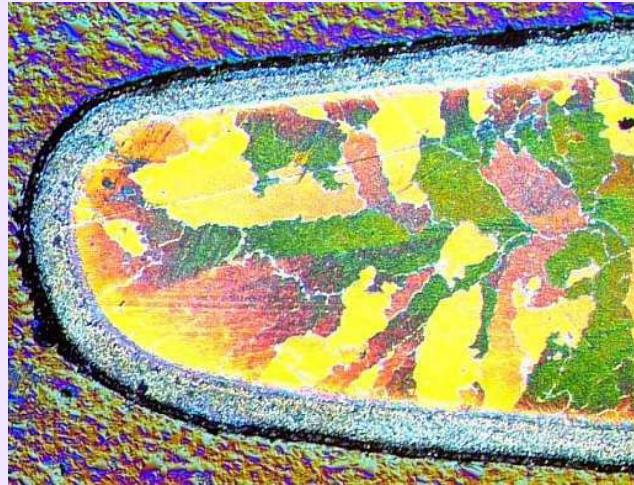
- This composition makes them highly suitable as casting metals
- Tonnage of cast iron castings is several times that of all other cast metal parts combined, excluding cast ingots in steel-making that are subsequently rolled into bars, plates, and similar stock
- Overall tonnage of cast iron is second only to steel among metals



## Question



What do you think this is?



Cross-section of a gray cast iron using an optical microscopy (up to 1000 times magnification)



## Types of Cast Irons



- Most important is gray cast iron
- Other types include ductile iron, white cast iron, malleable iron, and various alloy cast irons
- Ductile and malleable irons possess chemistries similar to the gray and white cast irons, respectively, but result from special processing treatments

Gray cast Iron → Special melting and pouring treatment (Chemical treatment) → Ductile Iron  
White cast Iron → Heat treatment → Malleable Iron



# Cast Iron Chemistries

Figure 6.13 Carbon and silicon % for cast irons, compared to steels (most steels have relatively low Si % - cast steels have higher Si %).

Ductile iron is formed by special melting and pouring treatment of gray cast iron.

Malleable iron is formed by heat treatment of white cast iron.

